

# Spectroscopic Characterization of the Soluble Guanylate Cyclase-like Heme Domains from *Vibrio cholerae* and *Thermoanaerobacter tengcongensis*<sup>†</sup>

David S. Karow,<sup>‡</sup> Duohai Pan,<sup>§,||</sup> Rosalie Tran,<sup>§</sup> Patricia Pellicena,<sup>⊥</sup> Andrew Presley,<sup>§</sup> Richard A. Mathies,<sup>§,¶</sup> and Michael A. Marletta<sup>\*,§,⊥,¶</sup>

Program in Cellular and Molecular Biology, University of Michigan, Ann Arbor, Michigan 48109, and Department of Chemistry, Department of Molecular and Cell Biology, and Division of Physical Biosciences, Lawrence Berkeley Laboratory, University of California, Berkeley, California 94720

Received March 31, 2004; Revised Manuscript Received May 28, 2004

**ABSTRACT:** Soluble guanylate cyclase (sGC) is a nitric oxide- (NO-) sensing hemoprotein that has been found in eukaryotes from *Drosophila* to humans. Prokaryotic proteins with significant homology to the heme domain of sGC have recently been identified through genomic analysis. Characterization of two of these proteins is reported here. The first is a 181 amino acid protein cloned from *Vibrio cholerae* (VCA0720) that is encoded in a histidine kinase-containing operon. The ferrous unligated form of VCA0720 is 5-coordinate, high-spin. The CO complex is low-spin, 6-coordinate, and the NO complex is high-spin and 5-coordinate. These ligand-binding properties are very similar to those of sGC. The second protein is the N-terminal 188 amino acids of Tar4 (*TtTar4H*), a predicted methyl-accepting chemotaxis protein (MCP) from the strict anaerobe *Thermoanaerobacter tengcongensis*. *TtTar4H* forms a low-spin, 6-coordinate ferrous-oxy complex, the first of this sGC-related family that binds O<sub>2</sub>. *TtTar4H* has ligand-binding properties similar to those of the heme-containing O<sub>2</sub> sensors such as A<sub>x</sub>PDEA1. sGC does not bind O<sub>2</sub> despite having a porphyrin with a histidyl ligand like the globins. The results reported here, with sequence-related proteins from prokaryotes but in the same family as the sGC heme domain, show that these proteins have evolved to discriminate between ligands such as NO and O<sub>2</sub>; hence, we term this family H-NOX domains (heme-nitric oxide/oxygen).

One important function of hemoproteins is to sense gaseous, diatomic small molecules (*1*). There are four distinct heme-binding domains that have been found in heme-based sensors. They can be segregated on the basis of their ligand specificity, functioning to sense either O<sub>2</sub>, CO, or NO. There are two domains that are specialized for O<sub>2</sub> sensing, both found in prokaryotes. The first is the Mb- (myoglobin-) like heme domain that is contained within HemAT (*2, 3*), an aerotaxis-regulating protein, and the second is the heme-PAS domain, which is found in the histidine kinase-containing protein, FixL (*4, 5*), and the phosphodiesterase-containing A<sub>x</sub>PDEA1 protein from *Acetobacter xylinum* (*6*). The third domain functions to sense CO and is found in the *Rhodospirillum rubrum* CoxA protein. CoxA regulates genes involved in CO oxidation (*7*). Finally, the last domain is specialized for NO binding and is found in the NO-

responsive soluble guanylate cyclase (sGC).<sup>1</sup> This enzyme, found in many animals, has a heme-containing N-terminus that binds NO, regulating a C-terminal guanylate cyclase (*8*). sGC catalyzes the conversion of GTP to cGMP and is a heterodimer, composed of  $\alpha$  and  $\beta$  subunits. Nitric oxide binds to the ferrous heme, which is ligated to the  $\beta$  subunit (*9*), and stimulates the activity of the enzyme over 200-fold (*10*). sGC has been characterized in eukaryotes ranging from *Drosophila* to humans; however, no sGCs or sGC-like proteins have been characterized in prokaryotes.

Recently, a PSI-BLAST search with the sGC heme domain as the query sequence identified multiple, predicted prokaryotic proteins with homology to this region (*11*); however, no further analysis was performed. We have now cloned and characterized two of these proteins: a 181 amino acid protein from *Vibrio cholerae* (VCA0720) and the heme domain (188 amino acids) from the *Thermoanaerobacter tengcongensis* Tar4 protein (*TtTar4H*). These proteins share spectroscopic properties, sequence homology, and key conserved residues with sGC, and therefore, we place them in the same family. Although conserved from prokaryotes to humans, the biological role for these sGC-like heme domains has likely

<sup>†</sup> This work was supported in part by LDRD funds from the Lawrence Berkeley National Laboratory (M.A.M.).

<sup>\*</sup> To whom correspondence should be addressed at the Department of Chemistry, University of California, Berkeley. Telephone: 510-643-9325. Fax: 510-643-9388. E-mail: marletta@berkeley.edu.

<sup>‡</sup> Program in Cellular and Molecular Biology, University of Michigan.

<sup>§</sup> Department of Chemistry, University of California, Berkeley.

<sup>||</sup> Present address: Pacific Northwest National Laboratory, K8-88, Richland, WA 99352.

<sup>⊥</sup> Department of Molecular and Cell Biology, University of California, Berkeley.

<sup>¶</sup> Division of Physical Biosciences, Lawrence Berkeley Laboratory, University of California, Berkeley.

<sup>1</sup> Abbreviations: sGC, soluble guanylate cyclase; *TtTar4H*, residues 1–188 of *Thermoanaerobacter tengcongensis* Tar4 (heme domain); MCP, methyl-accepting chemotaxis protein; A<sub>x</sub>PDEA1H, *Acetobacter xylinum* phosphodiesterase A1 heme domain; Mb, myoglobin; H-NOX domain, heme-nitric oxide/oxygen binding domain; DTT, dithiothreitol; DEA, diethylamine; TEA, triethylamine.

diverged, evidenced by the observation that it is contained in a histidine kinase containing operon in *V. cholerae*, a MCP in *T. tengcongensis*, and guanylate cyclases in eukaryotes.

Consistent with different functions, these proteins have very different ligand-binding properties. For example, as shown by resonance Raman and UV/vis spectroscopy, VCA0720 has ligand-binding characteristics essentially identical to those of sGC, while, strikingly, *TtTar4H* is able to bind O<sub>2</sub> and has ligand-binding characteristics most similar to those of Mb and known O<sub>2</sub> sensors such as FixL and A<sub>x</sub>PDEA1. These spectroscopic results suggest that with the appropriate amino acid substitutions the sGC-like heme domain can discriminate between NO and O<sub>2</sub>; hence, we term the proteins in this family H-NOX domains (*heme-nitric oxide/oxygen*).

## MATERIALS AND METHODS

**Construction of Expression Plasmids.** PCR was used to amplify *TtTar4H* [1–564 bp of TTE0680 (*Tar4*)] and VCA0720 from *T. tengcongensis* and *V. cholerae* genomic DNA (ATCC), respectively, using Expand polymerase (Roche). The upstream primers were 5'-ggaattccatgatgaagcattatctataccgttctc-3' and 5'-ggaattccatgaaggggacaatcgctgggacatgg-3' for VCA0720 and *TtTar4H*, respectively. The predicted VCA0720 protein sequence contained ten N-terminal amino acids that were not homologous to the other predicted H-NOX domains, and thus, the 5' end of the upstream primer corresponded to the Met residue that precedes Gln-Gly-Ile (M11). The downstream primers were 5'-atagtttagcggcgctcatgatggcaaaattccactac-3' and 5'-atagtttagcggcgctcaattttcttatactcaaaaacgggg-3' for VCA0720 and *TtTar4H*, respectively. Upstream and downstream primers (Qiagen) contained *Nde*I and *Not*I restriction sites, respectively. Amplified PCR products were cloned into pET-20b (Novagen) and sequenced (UC Berkeley sequencing core and Elim Biopharmaceuticals, Inc.).

**Escherichia coli Expression.** Expressions were performed as described previously (12) with the following modifications. Plasmids were transformed into Tuner DE3 plysS cells (Novagen). Cultures were grown until an OD<sub>600</sub> of 0.5–1 and cooled to 27 °C. Isopropyl β-D-thiogalactopyranoside (Promega) was added to 10 μM and aminolevulinic acid to 1 mM. Cultures were grown overnight for 14–18 h and then harvested.

**Protein Purification.** VCA0720 was purified in the following manner: Frozen cell pellets from 3 L of culture were thawed quickly at 37 °C and resuspended in 120 mL of buffer A [50 mM DEA, pH 8.5, 20 mM NaCl, 5 mM DTT, 1 mM Pefabloc (Pentapharm), and 5% glycerol]. Resuspended cells were lysed with sonication and, subsequently, with an Emulsiflex-C5 high-pressure homogenizer at 20000 psi (Avestin, Inc.). Lysed cells were centrifuged at 100000g for 40 min. The supernatant was applied to a 100 mL (2.5 × 21 cm) Toyopearl Q 650M anion-exchange column (Tosohas), washed with 3 column volumes of buffer A, and developed with a linear NaCl gradient from 20 to 500 mM (800 mL). The flow rate was 1.5 mL/min for loading and 1 mL/min for washing and eluting. Fractions were selected on the basis of the intensity of the red/brown color. The eluate (~50 mL) was concentrated to 4 mL using 15 mL 10K MWCO spin concentrators (Millipore). The concentrated proteins were

applied to a prepacked Superdex S200 HiLoad 26/60 gel filtration column (Pharmacia) that had been equilibrated with TEA, pH 7.5, 50 mM NaCl, 5mM DTT, and 5% glycerol. The flow rate was 1.3 mL/min. Fractions containing VCA0720 were pooled and applied to a POROS HQ 7.9 mL (1 × 10 cm, 10 μm) anion-exchange column (Applied Biosystems) at 10 mL/min that had been equilibrated with buffer B (50 mM DEA, pH 8.5, 50 mM NaCl, and 5% glycerol). The column was washed with 3 column volumes of buffer B at 10 mL/min and developed with a linear gradient of NaCl from 50 to 500 mM in a total volume of 140 mL at the same flow rate. VCA0720-containing fractions were pooled and stored at –70 °C.

*TtTar4H* was lysed in buffer A as described above for VCA0720. The supernatant was subjected to heat denaturation (75 °C for 40 min) and centrifuged at 100000g for 40 min. The supernatant was applied to the same Toyopearl Q column used for VCA0720. This column had been equilibrated with buffer A, and the flow-through was collected. The *TtTar4H*-containing flow-through was concentrated to 4 mL and subjected to gel filtration as described above for VCA0720 except that the buffer contained 50 mM TEA, pH 7.4, 200 mM NaCl, and 5% glycerol. *TtTar4H*-containing fractions were pooled and stored at –70 °C.

**UV/Vis Spectroscopy.** All spectra were recorded in an anaerobic cuvette on a Cary 3E spectrophotometer equipped with a Neslab RTE-100 temperature controller set at 10 °C. Spectra were recorded from proteins in a solution of 50 mM TEA, pH 7.5, and 50 mM NaCl (buffer C). The heme concentration was ~5 μM. Protein samples were placed into the anaerobic cuvette in an anaerobic glovebag (Coy) except for the Fe<sup>2+</sup>–O<sub>2</sub> *TtTar4H* complex. Fe<sup>2+</sup>-unligated protein samples were prepared in the following manner: Purified protein was made anaerobic in an O<sub>2</sub>-scavenged gas manifold with 10 cycles of alternate evacuation and purging with purified argon and brought into an anaerobic glovebag. For *TtTar4H*, ferricyanide (~100 equiv) was added to remove the bound O<sub>2</sub>. Ferricyanide was removed using a PD10 desalting column (Amersham Biosciences) that had been equilibrated with buffer C, and then the protein was reduced using dithionite (~100 equiv). The dithionite was then removed in the same manner. For consistency, VCA0720 was prepared in the same manner. The CO complexes were generated by adding Fe<sup>2+</sup>-unligated protein to a sealed Reacti-Vial (Pierce) that contained CO (Airgas). NO complexes were generated by preparing an anaerobic solution of diethylamine NONOate (Caymen) in buffer C (~10 mM) in a sealed Reacti-Vial. The headspace (1 mL) was then removed using a gastight syringe and delivered to an Fe<sup>2+</sup>-unligated protein sample contained in a sealed Reacti-Vial. Lastly, Fe<sup>2+</sup>–O<sub>2</sub> *TtTar4H* was formed by removing Fe<sup>2+</sup>-unligated *TtTar4H* from the anaerobic chamber and exposing to air.

**Extinction Coefficient Determination.** An anaerobic, Fe<sup>2+</sup>-unligated, UV/vis spectrum was recorded for both protein samples. These samples, along with the Mb standard, were prepared by oxidizing and rereducing each sample anaerobically as described above. The heme concentration was then determined for each sample by HPLC (13) and used to determine the Fe<sup>2+</sup>-unligated extinction coefficient. This extinction coefficient was then used to determine the heme concentration and extinction coefficients for the CO, NO,

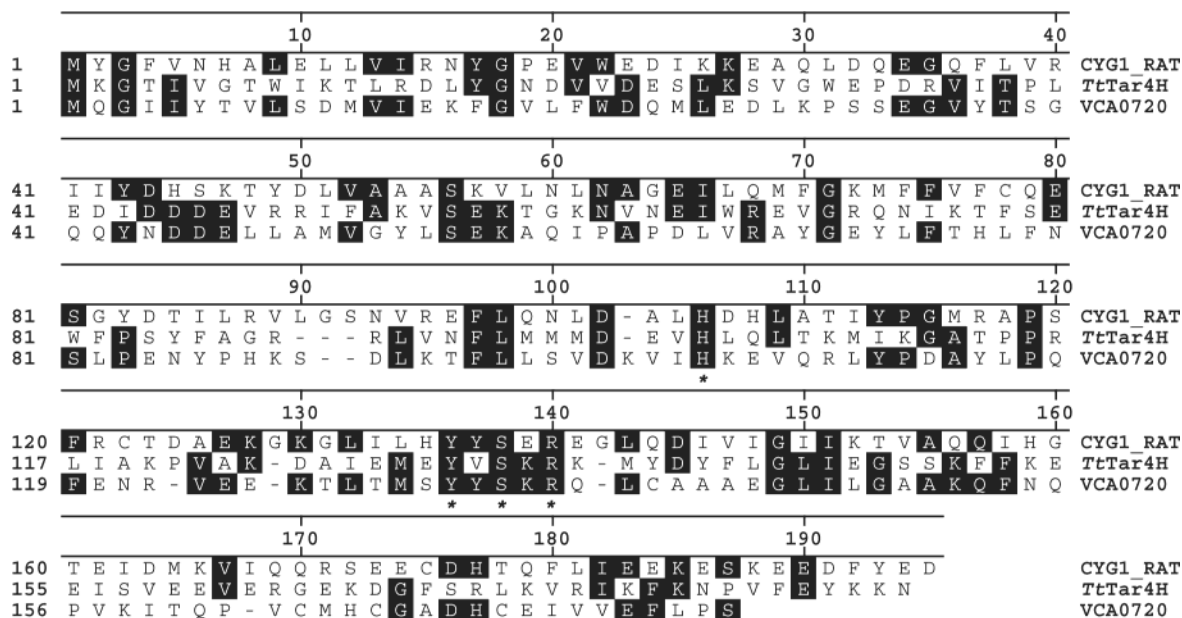


FIGURE 1: Alignment of rat  $\beta 1$ (1–194) (CYG1\_RAT) with *TtTar4H* and VCA0720. The alignment was generated using DNASTAR's Megalign program and the Clustal W algorithm. Conserved residues that are identical to the consensus sequence are shaded in black. Specific residues discussed in the text are marked with an asterisk (His105, the heme ligand, and Tyr135, Ser137, and Arg139 using the rat  $\beta 1$  numbering).

and  $\text{Fe}^{2+}$ – $\text{O}_2$  complexes from previously recorded UV/vis spectra. The HPLC method used to determine the heme concentration was performed with a protein C4 column (250  $\times$  4.6 mm, 5  $\mu\text{m}$ ; Vydac) and a Hewlett-Packard Series II 1090 HPLC with a diode array detector. Each sample (25  $\mu\text{L}$ ) was applied to the C4 column that had been equilibrated with 0.1% trifluoroacetic acid (TFA). The column was developed with a linear gradient of 0–75% acetonitrile over 15 min followed by a linear gradient of 75–95% acetonitrile over 5 min. The flow rate was 1 mL/min.

**Resonance Raman Spectroscopy.** Spectra of all samples were collected using 413.1 nm excitation from a krypton ion laser (Spectra-Physics 2025), except for the VCA0720 NO complex, which was excited at 406.7 nm. A microspinning sample cell was used to minimize photoinduced degradation. Raman scattering was detected with a cooled, back-illuminated CCD (LN/CCD-1100/PB; Roper Scientific) controlled by a ST-133 controller coupled to a subtractive dispersion, double spectrograph (14). Raman spectra were corrected for the wavelength dependence of the spectrometer efficiency by using a white lamp and calibrated with cyclohexane. The reported frequencies are accurate to  $\pm 2$   $\text{cm}^{-1}$ , and the resolution of the spectra is 8  $\text{cm}^{-1}$ . All Raman spectra were obtained by subtracting the buffer background and baseline correction. The laser power at the sample was 3 mW, except for the CO complexes where a power of 0.2 mW was used to avoid photolysis. Typical data acquisition times were 30 min. Absorption spectra taken pre- and post-Raman spectral collection confirmed the integrity of the ferrous unligated-*TtTar4H* and NO-*TtTar4H* samples. Approximate isotopic shifts were estimated on the basis of a simple diatomic harmonic oscillator model.

Protein samples were placed into the Raman cell in an anaerobic glovebag except for the  $\text{Fe}^{2+}$ – $^{16}\text{O}_2$  *TtTar4H* complex. The heme concentration was 17  $\mu\text{M}$  for VCA0720 and 47  $\mu\text{M}$  for *TtTar4H*. Protein samples for Raman spectroscopy were prepared as described above for UV/vis

spectroscopy with the following modifications.  $^{13}\text{CO}$  (99% containing  $\sim 10\%$   $^{18}\text{O}_2$ ; Cambridge Isotopes) and  $^{18}\text{O}_2$  (Spectra Gases) complexes were prepared in a manner similar to that of the  $^{12}\text{CO}$  complexes.  $^{15}\text{NO}$  complexes were made in the following manner: A 1 mL solution of 0.5 M  $\text{K}^{15}\text{NO}_2$  (Cambridge Isotopes) and a 1 mL solution of 1 M  $\text{KI}/\text{H}_2\text{SO}_4$  were made anaerobic on an oxygen-scavenging gas manifold and delivered into the anaerobic glovebag. A potassium nitrite solution (100  $\mu\text{L}$ ) was delivered to the KI solution using a gastight syringe.  $^{15}\text{NO}$ -containing headspace ( $\sim 1$  mL) was then delivered to a protein sample contained in a sealed Reacti-Vial.

## RESULTS

**Cloning, Expression, and Purification.** The predicted 181 amino acid ORF from *V. cholerae* (VCA0720) is encoded within a predicted histidine kinase-containing operon and is 22% identical to rat  $\beta 1$ (1–194). The N-terminal sGC-like heme domain region from *T. tengcongensis* was identified in the Tar4 protein, a predicted MCP. The heme domain of this protein was assumed to extend from residue 1 to residue 188, the last amino acid before the predicted membrane-spanning region. This sGC-like heme domain is 18% identical to rat  $\beta 1$ (1–194). An alignment of these two proteins with  $\beta 1$ (1–194) is shown in Figure 1. These two proteins were cloned and expressed in *E. coli* and purified as described in Materials and Methods. Purity was assessed by Coomassie staining of samples resolved by SDS-PAGE (data not shown). Except for minor contaminating bands, the proteins were pure. The yield was  $\sim 6$  and 20 mg/L for VCA0720 and *TtTar4H*, respectively. *TtTar4H* was isolated as the  $\text{Fe}^{2+}$ – $\text{O}_2$  complex (Soret maximum at 416 nm), and VCA0720 was isolated with a Soret maximum at 423 nm and a split  $\alpha/\beta$  region (data not shown). It was assumed that the VCA0720 sample was a mixture of ferrous and ferric oxidation states and was not further characterized. Upon reduction with dithionite, the isolated VCA0720 formed a



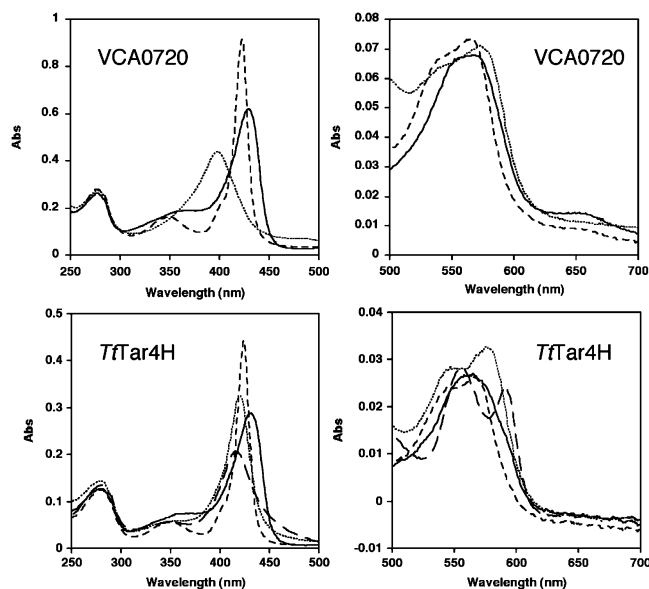


FIGURE 2: Electronic absorption spectra of VCA0720 and TrTar4H showing the Soret maximum and  $\alpha/\beta$  region: reduced, unligated (—); CO complex (---); NO complex (···); and O<sub>2</sub> complex (— · —). The heme concentration was  $\sim 5 \mu\text{M}$ .

Table 1: UV/Vis Peak Positions<sup>a</sup> and Extinction Coefficients<sup>b</sup>

protein	Soret	$\beta$	$\alpha$	ref
sGC	431 (111)	555 (14)		10
VCA0720	429 (137)	568 (15.0)		this work
TrTar4H	431 (124)	565 (11.6)		this work
Hb	430 (133)	555 (12.5)		15
CO bound				
sGC	423 (145)	541 (14)	567 (14)	10
VCA0720	423 (203)	541 (14.9)	566 (16.3)	this work
TrTar4H	424 (191)	544 (10.3)	565 (11.3)	this work
Hb	419 (191)	540 (13.4)	569 (13.4)	15
NO bound				
sGC	398 (79)	537 (12)	572 (12)	10
VCA0720	398 (97)	540 (14.1)	573 (15.7)	this work
TrTar4H	420 (140)	547 (12.2)	575 (14.0)	this work
Hb	418 (130)	545 (12.6)	575 (13.0)	15
O <sub>2</sub> bound				
TrTar4H	416 (89)	556 (12.2)	591 (10.4)	this work
Hb	415 (125)	541 (13.5)	576 (14.6)	15

<sup>a</sup> In nm. <sup>b</sup> In  $\text{mM}^{-1} \text{cm}^{-1}$ .

high-spin, 5-coordinate ferrous complex. UV/vis and resonance Raman spectroscopy were then used to characterize the heme environment and ligand-binding characteristics of both proteins.

**UV/Vis Absorption Spectroscopy.** UV/vis spectra are shown in Figure 2, and the data with extinction coefficients are summarized in Table 1 and compared to sGC (10) and hemoglobin (15).

**(A) Fe<sup>2+</sup>-Unligated and CO Complexes.** Both of these proteins have spectra that are similar to one another. The Fe<sup>2+</sup>-unligated spectra exhibit Soret bands at  $\sim 431 \text{ nm}$  and single, broad  $\alpha/\beta$  bands at  $\sim 560 \text{ nm}$ . Addition of CO to the anaerobically reduced proteins resulted in CO complexes with sharp Soret bands at  $\sim 424 \text{ nm}$ , increases in the extinction coefficient, and split  $\alpha/\beta$  bands. The shape and absorption maxima of the Soret and  $\alpha/\beta$  bands for both the Fe<sup>2+</sup>-unligated and CO complexes are very similar to those of full-length sGC, suggesting that the reduced proteins are 5-coordinate, high-spin and the CO complexes are 6-coordinate, low-spin.

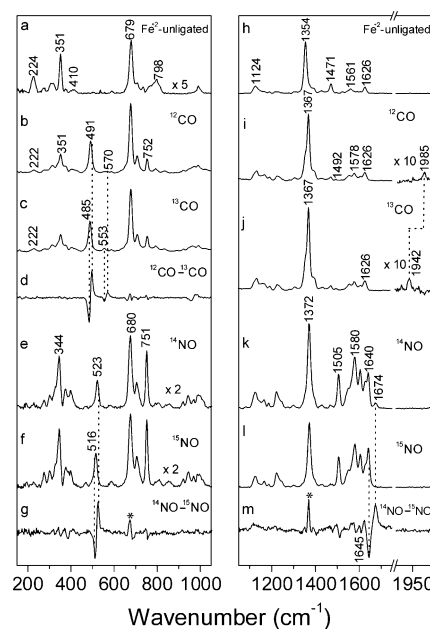


FIGURE 3: VCA0720 resonance Raman spectra: Fe<sup>2+</sup> unligated in the low-frequency (a) and high-frequency region (h); <sup>12</sup>CO and <sup>13</sup>CO complexes and their difference spectra in the low-frequency (b–d) and high-frequency regions (i and j), respectively; <sup>14</sup>NO and <sup>15</sup>NO complexes and their difference spectra in the low-frequency (e–g) and high-frequency regions (k–m), respectively. The heme concentration was  $17 \mu\text{M}$ . The subtraction factor in the isotopic difference spectra was adjusted to minimize features from vibrational modes that are insensitive to isotopic substitution. The asterisks indicate subtraction artifacts.

**(B) NO Complexes.** The NO complex of VCA0720 was also spectroscopically similar to sGC. Addition of NO to the Fe<sup>2+</sup>-unligated protein resulted in a spectrum with a Soret band at  $\sim 399 \text{ nm}$ , a decrease in the extinction coefficient, and a split  $\alpha/\beta$  band. This spectrum is similar to that of sGC and indicates a 5-coordinate, NO complex. However, for TrTar4H, the addition of NO resulted in a spectrum with a Soret band at  $420 \text{ nm}$  and a split  $\alpha/\beta$  band. This spectrum is similar to that of the 6-coordinate, NO complex from Hb (Table 1).

**(C) O<sub>2</sub> Complex.** TrTar4H forms a stable Fe<sup>2+</sup>–O<sub>2</sub> complex. Fe<sup>2+</sup>-unligated TrTar4H, contained in an anaerobic cuvette, was removed from the anaerobic chamber and exposed to air, and a spectrum was recorded. The spectrum exhibited a Soret band at  $416 \text{ nm}$  and a split  $\alpha/\beta$  band. This spectrum is similar to that of oxy-Hb and suggests that TrTar4H forms a low-spin, Fe<sup>2+</sup>–O<sub>2</sub> complex. The TrTar4H ferrous–oxy complex was very stable, exhibiting no spectroscopic changes overnight at  $4^\circ\text{C}$ .

**Resonance Raman Spectroscopy.** Raman spectra of VCA0720 and TrTar4H are shown in Figures 3 and 4, respectively, and the data are summarized in Table 2 along with comparisons to sGC (16, 17), Mb (18–23), and AxPDEA1H (the heme domain of AxPDEA1) (24).

**(A) Fe<sup>2+</sup>-Unligated Complexes.** In the high-frequency region, the heme skeletal marker bands for VCA0720 and TrTar4H are observed, including the  $1354 \text{ cm}^{-1}$   $\nu_4$  band that is sensitive to the heme oxidation state (25), the  $1471 \text{ cm}^{-1}$   $\nu_3$  band, and the  $\nu_2$  bands ( $1561$  and  $1575 \text{ cm}^{-1}$  for VCA0720 and TrTar4H, respectively). The  $\nu_3$  and  $\nu_2$  bands are sensitive to the spin and coordination state of the heme iron (25). The shoulder at  $1374 \text{ cm}^{-1}$  on the ferrous-unligated  $\nu_4$  peak at

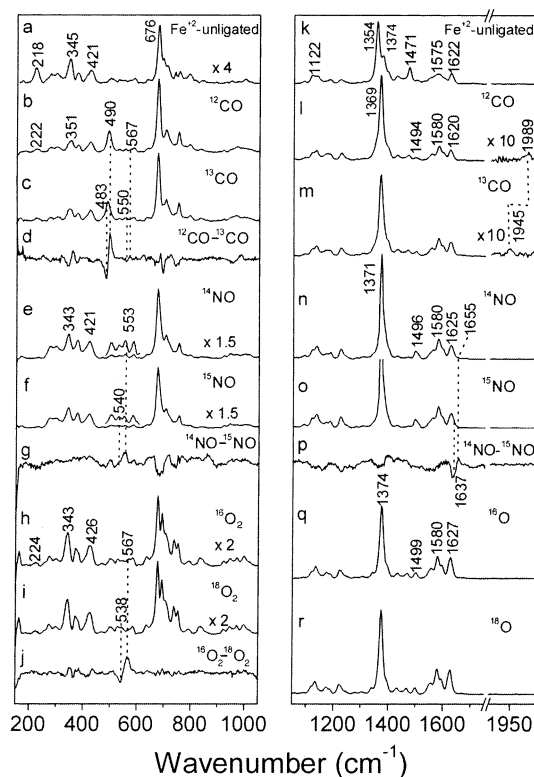


FIGURE 4: *TtTar4H* resonance Raman spectra:  $\text{Fe}^{2+}$  unligated in the low-frequency (a) and high-frequency region (k);  $^{12}\text{CO}$  and  $^{13}\text{CO}$  complexes and their difference spectra in the low-frequency (b–d) and high-frequency regions (l and m), respectively;  $^{14}\text{NO}$  and  $^{15}\text{NO}$  complexes and their difference spectra in the low-frequency (e–g) and high-frequency regions (n–p), respectively;  $^{16}\text{O}_2$  and  $^{18}\text{O}_2$  complexes and their difference spectra in the low-frequency (h–j) and high-frequency regions (q and r), respectively. The heme concentration was  $47 \mu\text{M}$ . The subtraction factor in the isotopic difference spectra was adjusted to minimize features from vibrational modes that are insensitive to isotopic substitution.

Table 2: Heme Skeletal Vibrations and Vibrational Modes<sup>a</sup>

protein	$\nu_2$	$\nu_3$	$\nu_4$	Fe–X	X–O	ref
<b><math>\text{Fe}^{2+}</math> unligated</b>						
sGC	1562	1471	1358	$\nu(\text{Fe–His})$		16, 17
VCA0720	1561	1471	1354	204		this work
<i>TtTar4H</i>	1575	1471	1354	224		this work
Mb	1563	1471	1357	218		18–23
AxPDEA1H	1557	1469	1354	220		24
<b>CO bound</b>						
sGC	1583	1500	1371	$\nu(\text{Fe–CO})$	$\nu(\text{C–O})$	16, 17
VCA0720	1578	1492	1367	472	1987	this work
<i>TtTar4H</i>	1580	1494	1369	491	1985	this work
Mb	1587	1498	1372	490	1989	18–23
AxPDEA1H	nr <sup>b</sup>	nr	nr	512	1944	24
<b>NO bound</b>						
sGC	1584	1509	1375	$\nu(\text{Fe–NO})$	$\nu(\text{N–O})$	16, 17
VCA0720	1580	1505	1372	525	1677	this work
<i>TtTar4H</i>	1580	1496	1371	523	1674	this work
Mb	1584	1501	1375	553	1655	18–23
AxPDEA1H	nr	nr	nr	554	1624	24
<b><math>\text{O}_2</math> bound</b>						
<i>TtTar4H</i>	1580	1499	1374	$\nu(\text{Fe–O})$		this work
Mb	1584	1507	1377	567		18–23
AxPDEA1H	nr	nr	nr	570		24

<sup>a</sup> In  $\text{cm}^{-1}$ . <sup>b</sup> Not reported.

$1354 \text{ cm}^{-1}$  was due to incomplete exclusion of oxygen from the sample during the course of the experiment. However, the  $\nu_3$  band at  $1471 \text{ cm}^{-1}$  indicates ferrous unligated protein

as the main sample component. The heme skeletal markers are similar to the corresponding bands in histidine-ligated, 5-coordinate, high-spin,  $\text{Fe}^{2+}$  hemoproteins such as sGC (Table 2). In the low-frequency spectrum, bands at 224 and  $218 \text{ cm}^{-1}$  are observed for VCA0720 and *TtTar4H*, respectively. These bands are assigned to the  $\nu(\text{Fe–His})$  stretching mode on the basis of previous observations that, in most histidine-ligated,  $\text{Fe}^{2+}$ –heme complexes, this stretching frequency is usually between 200 and  $250 \text{ cm}^{-1}$  (26), and that, in the NO and CO adducts, these bands are absent. On the basis of the above, both reduced proteins contain a 5-coordinate, high-spin heme.

**(B) CO Complexes.** The heme skeletal markers at 1369, 1494, and  $1580 \text{ cm}^{-1}$  are similar to the corresponding bands from histidine-ligated, 6-coordinate, low-spin,  $\text{Fe}^{2+}$  hemoproteins such as sGC (Table 2). In the low-frequency spectrum, isotope-sensitive bands at 491 and  $490 \text{ cm}^{-1}$  are observed for VCA0720 and *TtTar4H*, respectively. The predicted values for  $^{13}\text{CO}$  isotopic frequencies are  $485 \text{ cm}^{-1}$  for VCA0720 and  $484 \text{ cm}^{-1}$  for *TtTar4H*. The observed frequencies of 485 and  $483 \text{ cm}^{-1}$ , respectively, are in excellent agreement. These bands are assigned to the  $\nu(\text{Fe–CO})$  stretching mode on the basis of their isotopic shift and their similarity to the reported  $\nu(\text{Fe–CO})$  frequency for AxPDEA1H ( $493 \text{ cm}^{-1}$ ). In the high-frequency spectrum we observed isotope-sensitive bands at 1985 and  $1989 \text{ cm}^{-1}$  for VCA0720 and *TtTar4H*, respectively. The isotopic shifts also agree well with calculated values (calculated, 1941 and  $1945 \text{ cm}^{-1}$ , and actual, 1942 and  $1945 \text{ cm}^{-1}$ , for VCA0720 and *TtTar4H*, respectively). These bands are assigned to the  $\nu(\text{CO})$  stretching mode on the basis of their isotopic shift and their similarity to the reported  $\nu(\text{CO})$  frequency of  $1987 \text{ cm}^{-1}$  for full-length sGC as determined by FTIR. The CO isotopic sensitive vibration band at  $979 \text{ cm}^{-1}$  can be assigned to the first overtone of the  $\nu(\text{Fe–CO})$  stretching vibration. Like sGC, VCA0720 and *TtTar4H* have high  $\nu(\text{CO})$  stretching frequencies, and the  $1989 \text{ cm}^{-1}$  value observed for *TtTar4H* is the highest yet reported. Therefore, both proteins form CO complexes that are most similar to sGC: These complexes are 6-coordinate, low-spin and are notable for their high  $\nu(\text{CO})$  stretching frequency.

**(C) NO Complexes.** For the NO complex of VCA0720 (Figure 3), the high-frequency spectrum shows heme skeletal marker bands that are similar to the corresponding bands in 5-coordinate NO-sGC (Table 2). In the low-frequency spectrum, an isotope-sensitive band is observed at  $523 \text{ cm}^{-1}$ . The estimated  $^{15}\text{NO}$  isotopic shift of about  $7 \text{ cm}^{-1}$  agrees well with the observed  $\nu(\text{Fe–NO})$  shift from 523 to  $517 \text{ cm}^{-1}$ . This band is assigned to the  $\nu(\text{Fe–NO})$  stretching mode on the basis of its isotopic shift and similarity to the  $\nu(\text{Fe–NO})$  frequency in sGC ( $525 \text{ cm}^{-1}$ ). We also observed an isotope-sensitive band in the high-frequency spectrum ( $1674 \text{ cm}^{-1}$ ). We estimate an isotopic shift value for  $^{15}\text{N–O}$  stretching of about  $27 \text{ cm}^{-1}$ , which agrees well with our observation that the N–O stretch shifts from 1674 to  $1645 \text{ cm}^{-1}$ . On the basis of its isotopic shift and similarity to the  $\nu(\text{NO})$  frequency in sGC ( $1677 \text{ cm}^{-1}$ ), this band is assigned to the  $\nu(\text{NO})$  stretching mode. The NO complex of VCA0720 is, therefore, 5-coordinate and similar to sGC.

For *TtTar4H* (Figure 4), the high-frequency spectrum exhibits heme skeletal marker bands that are similar to the corresponding bands in 6-coordinate, low-spin NO-Mb. In

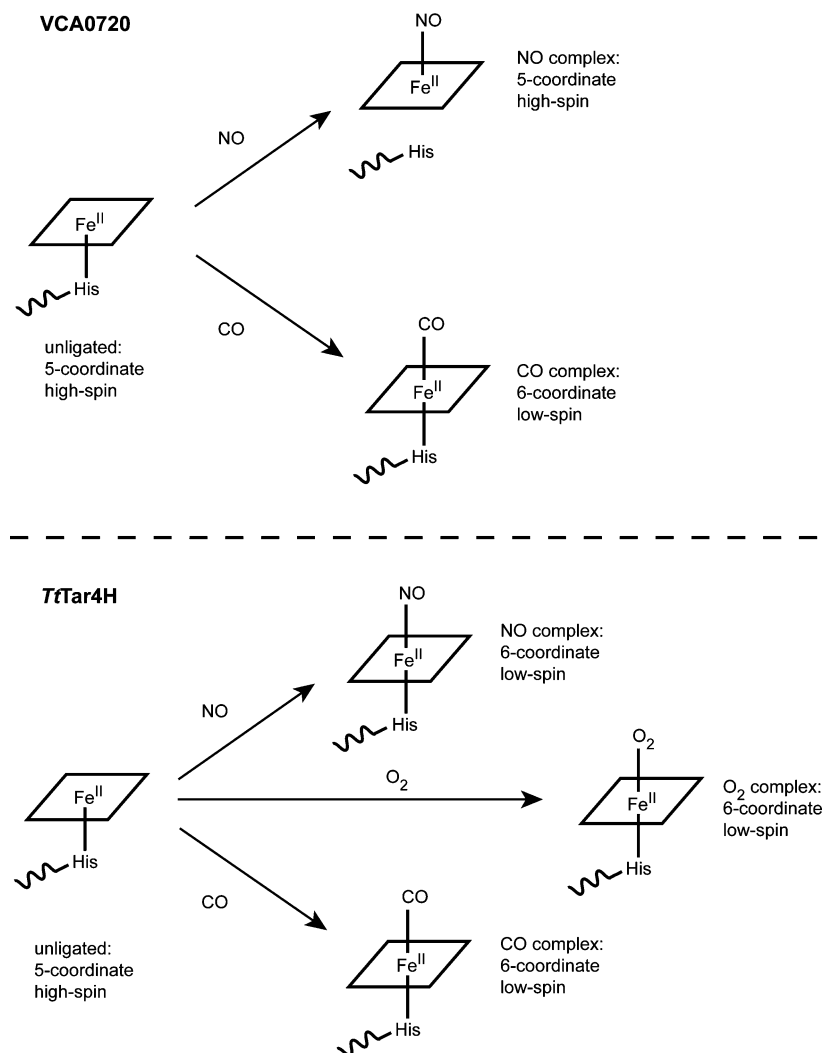


FIGURE 5: Summary of the results with VCA0720 and *TtTar4H*. A summary of the spectroscopic results is shown. VCA0720 forms stable complexes with NO and CO but not with O<sub>2</sub> and therefore has ligand-binding properties similar to those of the NO receptor soluble guanylate cyclase. *TtTar4H*, on the other hand, is isolated as a ferrous-oxy complex and has spectral properties similar to those of the globin family and, like the globins, forms stable complexes with NO and CO.

the low-frequency spectrum, bands in the 500–600 cm<sup>-1</sup> region are weakly sensitive to isotopic substitution of the NO ligand. In particular, the 553 cm<sup>-1</sup> band is plausibly attributed as the  $\nu(\text{Fe}-\text{NO})$  stretching mode based on its similarity to 6-coordinate Mb and NO-AxPDEA1H (554 and 560 cm<sup>-1</sup>, respectively). This band exhibits its sensitivity to <sup>14</sup>NO  $\rightarrow$  <sup>15</sup>NO isotopic substitution, but the S/N ratio in the difference spectrum is insufficient to determine the shifted frequency. We estimate a  $\nu(\text{Fe}-\text{NO})$  frequency shift of 15 cm<sup>-1</sup> upon <sup>15</sup>NO substitution. In the high-frequency spectrum, an isotope-sensitive band at 1655 cm<sup>-1</sup> is observed. This band is assigned to the  $\nu(\text{NO})$  stretching mode on the basis of its isotopic shift (1637 cm<sup>-1</sup>; calculated value 1625 cm<sup>-1</sup>) and similarity to 6-coordinate AxPDEA1H (1637 cm<sup>-1</sup>). We conclude that *TtTar4H* forms a 6-coordinate, low-spin NO complex that is similar to AxPDEA1H.

(D) *O<sub>2</sub> Complex*. For *TtTar4H* (Figure 4), the high-frequency spectrum exhibits heme skeletal marker bands that are similar to the corresponding bands in 6-coordinate, low-spin O<sub>2</sub>-Mb. In the low-frequency spectrum, a single isotope-sensitive band at 567 cm<sup>-1</sup> is observed. The predicted isotopic shift for the <sup>18</sup>O<sub>2</sub>-*TtTar4H* complex is 25 cm<sup>-1</sup>, and the experimental value was 29 cm<sup>-1</sup>. This band is assigned

to the  $\nu(\text{Fe}-\text{O}_2)$  stretching mode on the basis of its isotopic shift and similarity to the  $\nu(\text{Fe}-\text{O}_2)$  frequency in AxPDEA1H (567 cm<sup>-1</sup>). *TtTar4H* forms an O<sub>2</sub> complex that is similar to AxPDEA1H and is 6-coordinate and low-spin.

## DISCUSSION

*VCA0720 and TtTar4H Are Related to the Heme Domain of sGC*. We assign VCA0720 and *TtTar4H* to the same family as the heme domain of sGC on the basis of the following similarities: First, both proteins have sequence homology to the heme domain of sGC (~20% identity). Second, sequence alignments with sGC reveal that specialized residues are conserved. For example, when compared to the rat  $\beta$ 1 subunit, we find that His105 and a C-terminal Tyr-X-Ser-X-Arg (YSR) motif are conserved. His105 is the protein ligand to the heme cofactor in sGC, and a recent study has shown that the tyrosine and the arginine from the YSR motif are critical for heme binding (27). Finally, both of these proteins share spectroscopic properties with sGC. Like sGC, the Fe<sup>2+</sup>-unligated forms of these proteins are 5-coordinate, high-spin, and the CO complexes are 6-coordinate, low-spin (Figure 5).



**Divergent Ligand-Binding Properties.** Although VCA0720, *TtTar4H*, and sGC have some common characteristics mentioned above, there are also some significant differences. First, we find different coordination states for NO: *TtTar4H* forms a 6-coordinate, low-spin NO complex like Mb while VCA0720 forms a 5-coordinate complex similar to sGC. Second, we find different reactivity toward oxygen: *TtTar4H* is able to form a stable ferrous-oxy complex while VCA0720, like sGC, does not. There are also more subtle differences as well. For example, although VCA0720 and *TtTar4H* have similar  $\nu(\text{CO})$  frequencies compared to sGC, their  $\nu(\text{Fe}-\text{CO})$  frequencies are moderately higher (491 and 490  $\text{cm}^{-1}$  for VCA0720 and *TtTar4H*, respectively, compared to 472  $\text{cm}^{-1}$  for sGC). sGC has one of the highest  $\nu(\text{CO})$  and lowest  $\nu(\text{Fe}-\text{CO})$  stretching frequencies, and on the basis of this observation, we have previously suggested that the distal pocket of sGC has significant negative polarity (28). This is based on the observation that the  $\nu(\text{CO})$  and  $\nu(\text{Fe}-\text{CO})$  frequencies are inversely correlated and are sensitive to distal pocket polarity (29, 30). Negative polarity in the distal pocket can inhibit  $\text{Fe } d_{\pi} \rightarrow \text{CO } \pi^*$  back-bonding, weakening the Fe-CO bond order and strengthening the C-O bond order. The moderately higher  $\nu(\text{Fe}-\text{CO})$  values for VCA0720 and *TtTar4H* suggest that their distal pockets may be different from sGC, possibly containing a hydrophobic environment rather than a negative one.

Lastly, we find varying  $\nu(\text{Fe}-\text{His})$  stretching frequencies (Table 2). VCA0720 and *TtTar4H* have higher frequencies that are closer to Mb (220  $\text{cm}^{-1}$ ) while sGC has an unusually low value (204  $\text{cm}^{-1}$ ). The  $\nu(\text{Fe}-\text{His})$  frequency is a direct measure of the Fe-His bond strength and is influenced by the proximal pocket environment (26). A weak bond is usually correlated with either the absence of a hydrogen bond to the histidine N $\delta$  proton (less imidazolate character) or strain on the Fe-His bond that is induced by the protein. Both have been implicated in the weak Fe-His bond in sGC (16, 31). The weak Fe-His bond in sGC (204  $\text{cm}^{-1}$ ) has also been implicated in its inability to form a stable ferrous-oxy complex (32), reflecting decreased electron density in the Fe  $d_{\pi}$  orbitals and less back-bonding to the O $_2$   $\pi^*$  orbitals. The ability of *TtTar4H* to form a stable ferrous-oxy complex is consistent with its higher  $\nu(\text{Fe}-\text{His})$  frequency (218  $\text{cm}^{-1}$ ), which is similar to myoglobin (220  $\text{cm}^{-1}$ ). However, VCA0720 has a much stronger  $\nu(\text{Fe}-\text{His})$  frequency (224  $\text{cm}^{-1}$ ), and yet it does not bind oxygen, suggesting that, for VCA0720, distal effects may be an important determinant in the ability to exclude the formation of a stable ferrous-oxy complex.

It is likely that subtle differences in the proximal and distal pockets will be responsible for the variability seen in the Fe-His bond strengths,  $\nu(\text{Fe}-\text{CO})$  frequencies, NO coordination complexes, and ferrous-oxy complexes in the sGC-like heme domain family. This variation is consistent with a recent study of four different members of the heme-PAS domain family where significant differences in their heme coordination states,  $\nu(\text{Fe}-\text{His})$  frequencies, ligand-binding properties, and stretching frequencies were reported (24). This study, along with our findings here, raises an important question: If individual members of a heme domain family show significant variability in their ligand-binding properties and apparent function, then which features, if any, are invariant in individual family members; i.e., are there

components that are completely conserved in every family member regardless of its individual function and ligand specificity? Sequence analysis of the sGC-like heme domain-containing family members reveals a conserved YSR motif that has been shown to be important for heme binding in sGC (27). We hypothesize that this motif will be part of a conserved core that is necessary for the function of this domain in all family members regardless of individual variations such as whether they bind O $_2$  or NO.

**sGC-like Heme Domains That Discriminate between NO and O $_2$ .** What might our spectroscopic findings predict about the function of these prokaryotic proteins? Our data show that VCA0720 is similar to sGC, forming a 5-coordinate NO complex. In fact, the  $\nu(\text{NO})$  and  $\nu(\text{Fe}-\text{NO})$  stretching frequencies are almost exactly the same as sGC, indicating a very similar NO-bound pocket. In sGC, the formation of a 5-coordinate NO complex and the breaking of the Fe-His bond are thought to play a central role in activation of this enzyme (33). Furthermore, like sGC, VCA0720 is unable to form a stable  $\text{Fe}^{2+}-\text{O}_2$  complex. sGC functions in an aerobic environment ( $\sim 50 \mu\text{M O}_2$ ) and has evolved to exclude oxygen binding in order to sense NO, which is present at much lower concentrations ( $\sim 1 \text{ nM}$ ) (34, 35). Thus, given these spectral similarities, it is likely that VCA0720 also acts as an NO sensor. VCA0720 and its prokaryotic orthologues are encoded in histidine kinase-containing operons. It is possible that VCA0720 binds NO and this NO-VCA0720 complex then regulates the operon-associated histidine kinase. NO, we speculate, could be derived from using NO $_3^-$  as an electron acceptor when O $_2$  levels are low. NO then would serve as a signal for growth under low O $_2$  tension and could then regulate genes required for growth under these conditions using this operon.

*TtTar4H*, on the other hand, does form a stable  $\text{Fe}^{2+}-\text{O}_2$  complex, which is similar to heme-bound O $_2$  sensors such as AxPDEA1. This is the first report of the formation of a stable  $\text{Fe}^{2+}-\text{O}_2$  complex in any protein from the sGC-like heme domain family. Because *T. tengcongensis* is a strict anaerobe, it seems likely that *TtTar4H* serves as an O $_2$  sensor. The *TtTar4H* heme domain comprises residues 1–188 of a predicted 602 amino acid protein. In the predicted protein, this heme domain is followed by a membrane-spanning region and then a methyl-accepting chemotaxis domain. We speculate that methylation of the MCP in response to O $_2$  binding serves as an initiating event in chemotaxis. It is tempting to speculate that *TtTar4H* and its orthologues, found only in strict anaerobes, function to sense O $_2$  and avoid this toxic environment.

The  $\nu(\text{Fe}-\text{O})$  frequency of *TtTar4H* at 567  $\text{cm}^{-1}$  is close to that seen in Mb (570  $\text{cm}^{-1}$ ). In globins, differences in the  $\nu(\text{Fe}-\text{O})$  stretching frequencies have been attributed to differences in hydrogen bonding to the oxygen ligand.  $\nu(\text{Fe}-\text{O})$  frequencies in the region of 570  $\text{cm}^{-1}$  are associated with stabilizing hydrogen bonds to the distal oxygen atom (36). The *TtTar4H*  $\nu(\text{Fe}-\text{O})$  frequency of 567  $\text{cm}^{-1}$  is consistent with a stabilizing hydrogen bond that could enhance oxygen affinity (36–38).

In summary, VCA0720 and *TtTar4H* are hemoproteins that are related to the heme domain from sGC and are able to discriminate between NO and O $_2$ . This conserved domain was previously named the HNOB domain (for heme nitric oxide binding domain), based solely on its sequence homol-

ogy to NO-sensing sGC (11). Our data suggest that this name does not adequately reflect the divergent ligand-binding properties of this domain and we, therefore, propose that this domain be named the H–NOX (heme–nitric oxide/oxygen binding) domain. Further analysis of the H–NOX domain is currently underway in order to understand molecular aspects of ligand discrimination and to establish its biological function.

## ACKNOWLEDGMENT

We thank Chinmay Majmudar for help with the initial characterization of  $\beta 1$ (1–194) and Dr. Elizabeth Boon for critical input and experimental expertise.

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BI049374L